**Interactive comment on “Detection of HO\textsubscript{2} by laser-induced fluorescence: calibration and interferences from RO\textsubscript{2} radicals” by H. Fuchs et al.**

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In further support of the findings presented in the paper and the results discussed in the short comment posted by Dillon, recent experiments conducted in Leeds have also shown high HO\textsubscript{2} yields from a range of RO\textsubscript{2} species reacting with NO. The tests were conducted using the University of Leeds OH reactivity instrument. A small fluorescence cell (held at \(\sim1\) Torr) sampled from a reduced pressure (\(\sim400\) Torr) stainless steel flow-tube. A mixture of ozone, humidified nitrogen, oxygen and hydrocarbon were introduced into the flow tube. The ozone was photolysed by a 266 nm laser pulse to generate O1D which subsequently reacted with H2O vapour to generate OH. The OH then reacted with the hydrocarbon present to form an RO\textsubscript{2} radical. Upon addition of NO to the fluorescence cell, close to 100% yield of HO\textsubscript{2} was observed from ethene-
derived RO2 radicals and \(\sim 90\%\) yield from isoprene-derived RO2. Significant HO2 yields were also observed for the longer chain (>C3) alkane-derived RO2 radicals: A \(\sim 17\%\) yield of HO2 from n-butane and \(\sim 44\%\) yield from n-pentane. In the most part these yields are consistent with the Master Chemical Mechanism predictions (although the MCM predicts a HO2 yield closer to 70% for n-pentane). In the fluorescence cell used for these experiments the NO was injected into the centre of the ambient air-stream via 1/8” stainless steel tubing, this led to almost complete conversion of HO2 to OH upon addition of \(\sim 5\) sccm NO. Significantly reducing the amount of NO added did not provide distinction between RO2 radicals and HO2. Varying the pinhole size from 1 mm to 0.2 mm also had little effect, suggesting that the conversion reaction time within this particular fluorescence cell (estimated to be \(\sim 6\) ms from these tests) was not significantly reduced by this change.

Tests have also been conducted on the University of Leeds ground-based FAGE instrument which uses much larger fluorescence cells. Under the operating conditions employed during the 2008 OP3 campaign that took place in the Borneo rainforest we have found that this instrument was relatively insensitive to detection of these RO2 species. During the campaign only one fluorescence cell was used to make alternate measurements of OH and HO2 and conditions were optimised to maximise its sensitivity to OH. Under this configuration incomplete mixing of NO into the ambient air-stream for the HO2 titration was evident and resulted in low conversion of HO2 to OH. Coupled with the high pumping capacity and therefore fast gas throughput of the fluorescence cell this configuration effectively minimised the potential interference.